

Sulfoxide-Promoted Functionalization of Allylic C-H Bond

Zeng Yuwen

2014-2-24

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2 Original Reaction and Its Mechanism

3 Reaction Evolution and Derivation

4 Synthetic Applications

Contents



1 Introduction

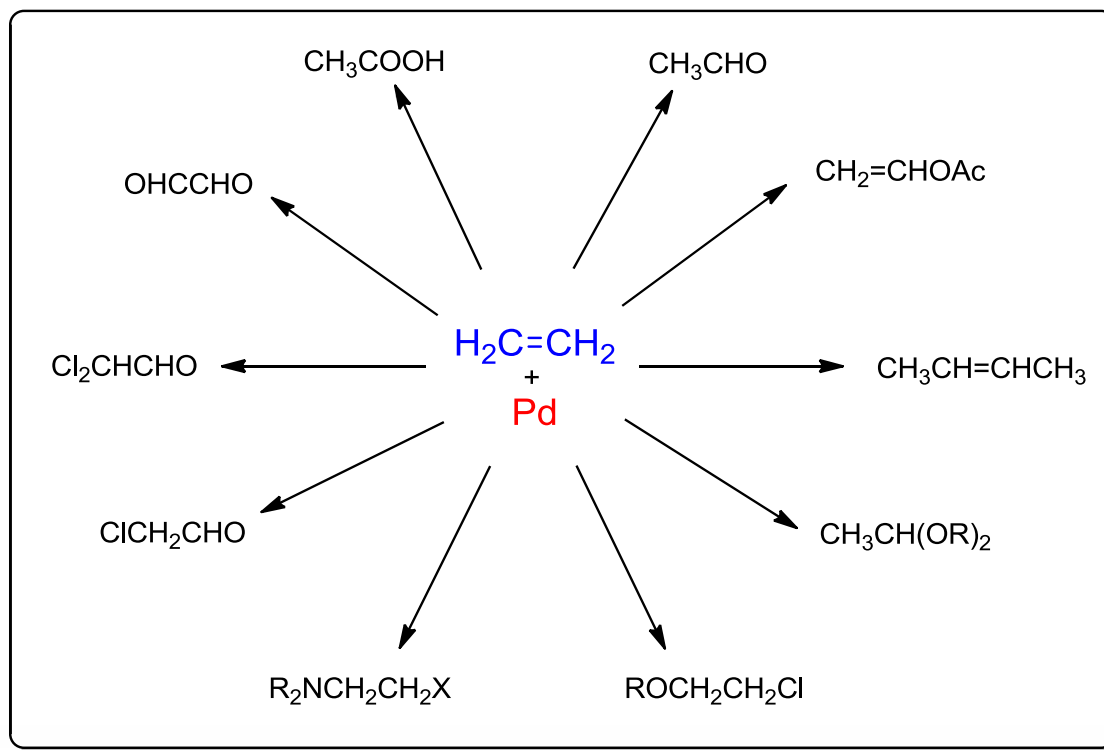
2 Original Reaction and Its Mechanism

3 Reaction Evolution and Derivation

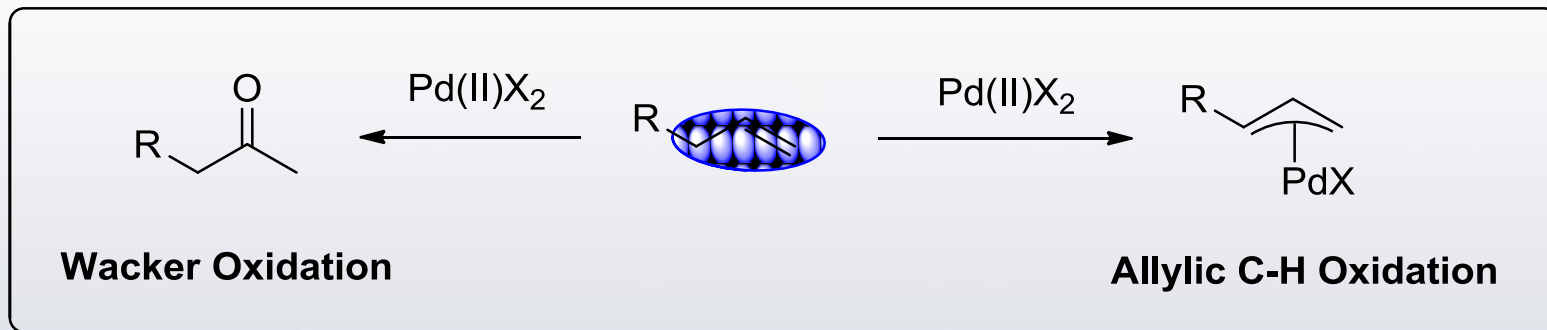
4 Synthetic Applications

Introduction

Palladium-Mediated Transformation of Olefins

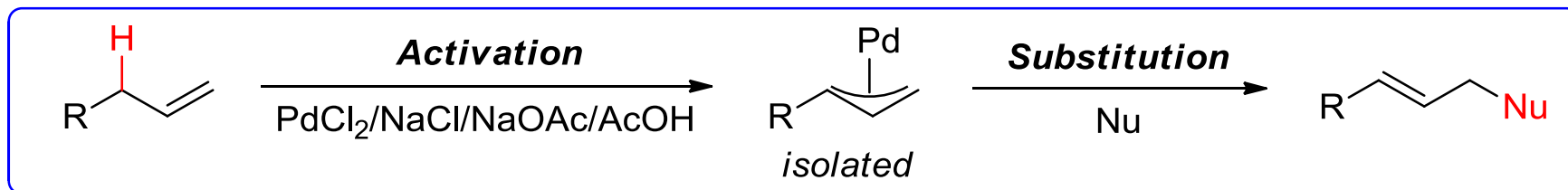


Palladium-Catalyzed Selective Oxidation of Olefins



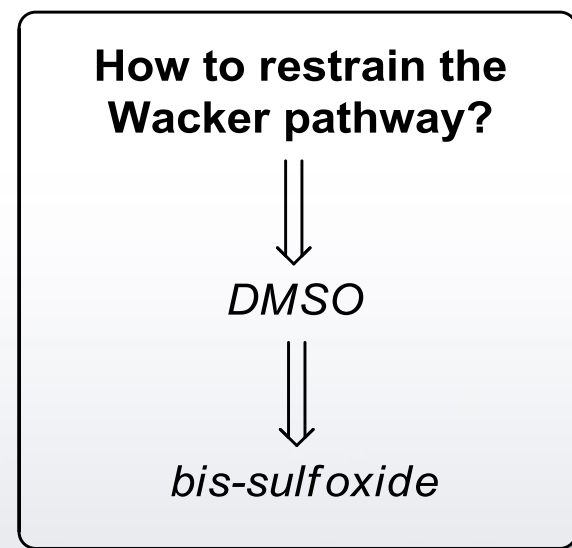
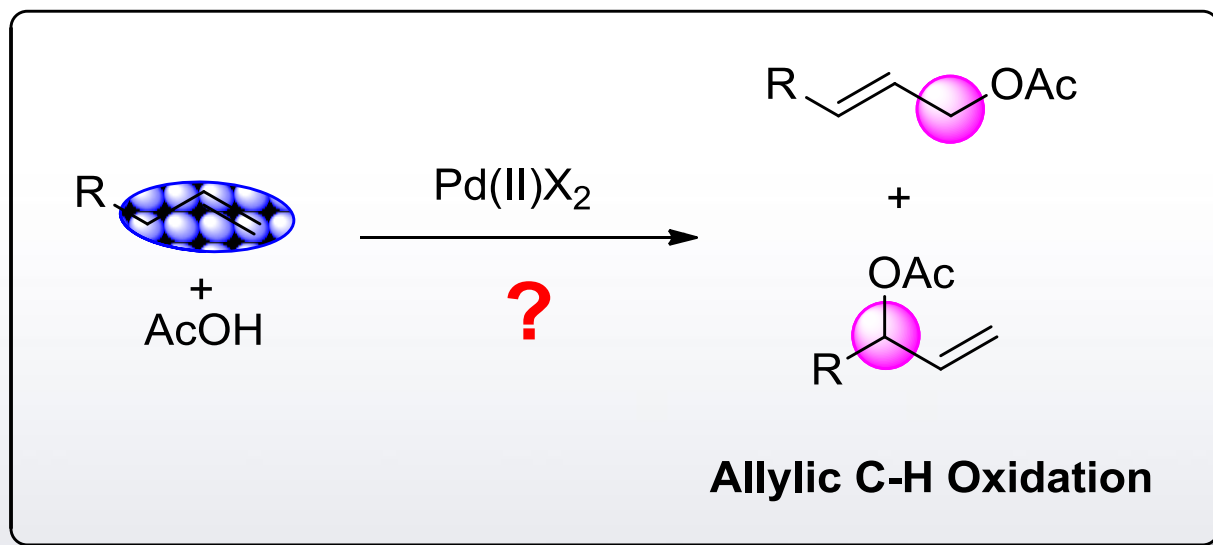
Introduction

Stoichiometric Palladium-Mediated, Two-Step Allylic C-H Alkylation



Huttel, R., Christ, H. *Chem. Ber.* **1964**, *97*, 2710.
Huttel, R.; Mcniff, M. *Chem. Ber.* **1973**, *106*, 1789.

Palladium-Catalyzed, One-Pot Allylic C-H Functionalization



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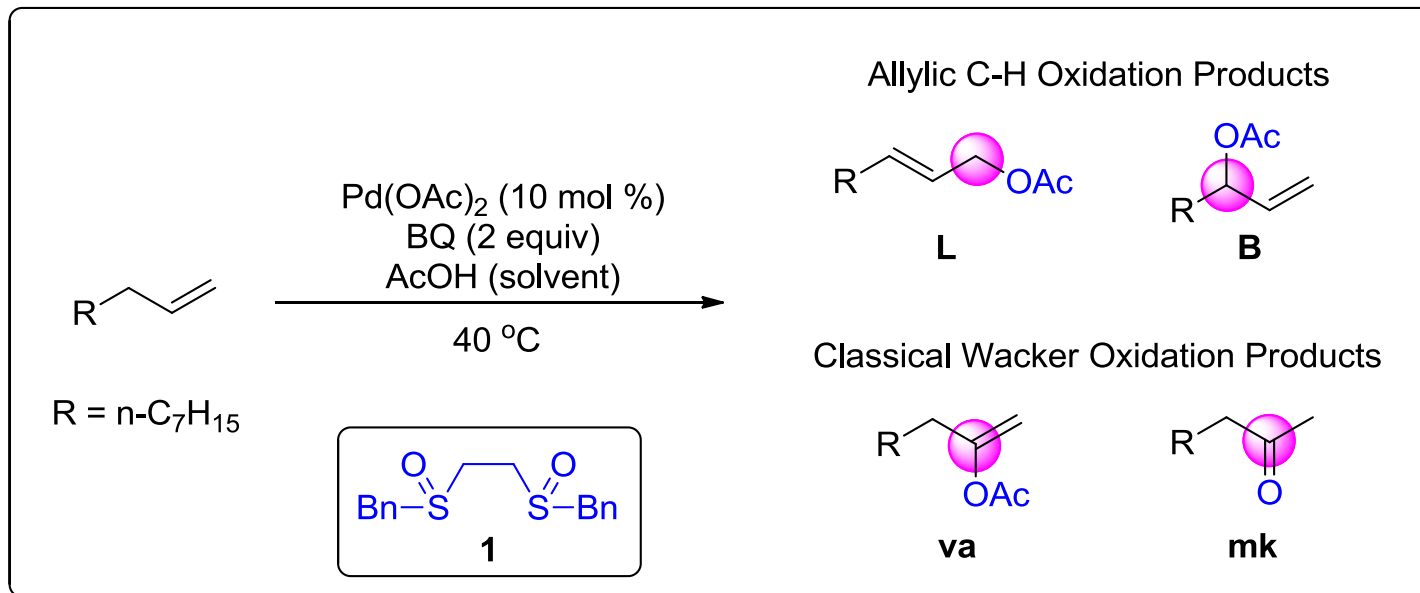


2 Original Reaction and Its Mechanism

3 Reaction Evolution and Derivation

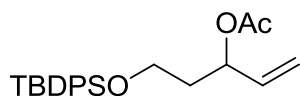
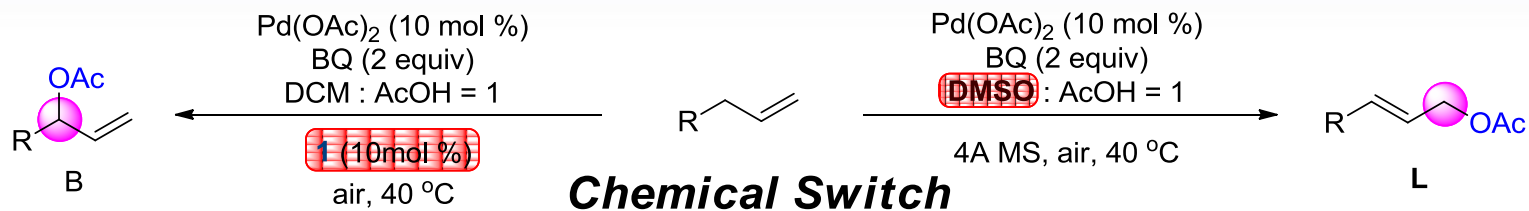
4 Synthetic Applications

Original Reaction

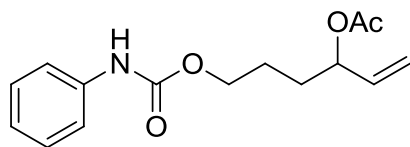


entry	additive	% yield (GC), 48 h				[L:B]
		L	B	va	mk	
1	none	3	5	17	14	1:2
2	DMSO (50 % V)	40	2	3	6	20:1
3	1 (10 mol %) DCM:AcOH (1:1)	8	66	<1	<1	1:8

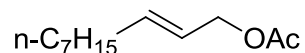
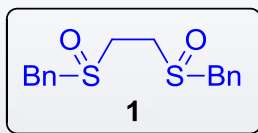
Original Reaction



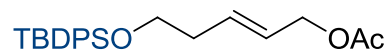
65 %, [L:B] = 1:6



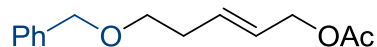
69 %, [L:B] = 1:5



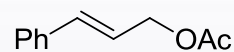
52 %, [L:B] = 24:1



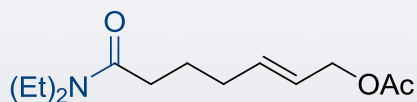
50 %, [L:B] = 31:1



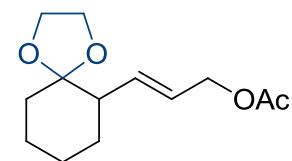
57 %, [L:B] = 31:1



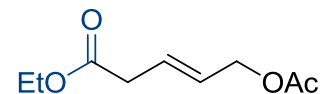
65 %, [L:B] > 99:1



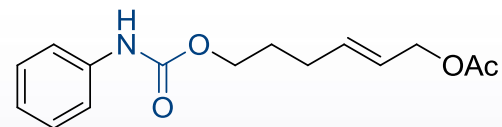
62 %, [L:B] = 23:1



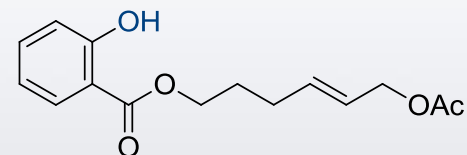
50 %, [L:B] > 99:1



54 %, [L:B] > 20:1

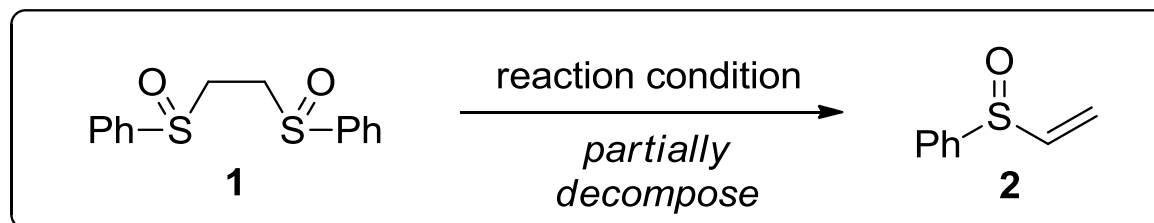


64 %, [L:B] = 13:1

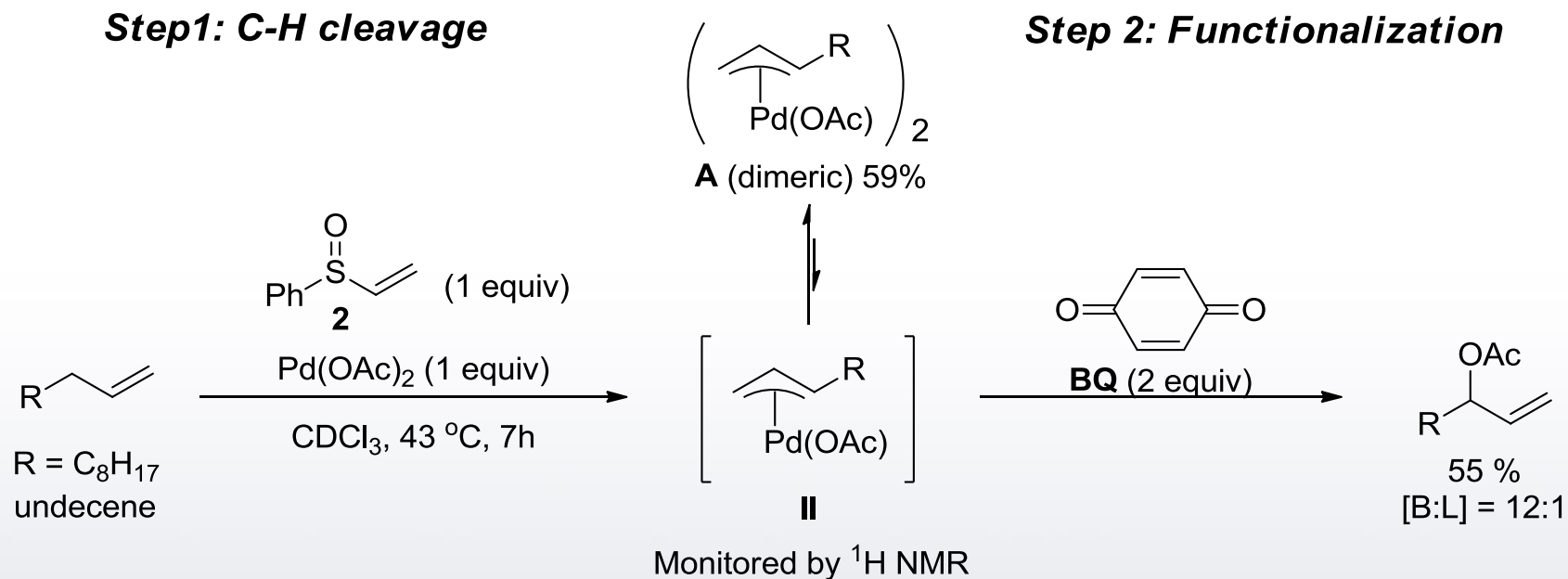


61 %, [L:B] = 14:1

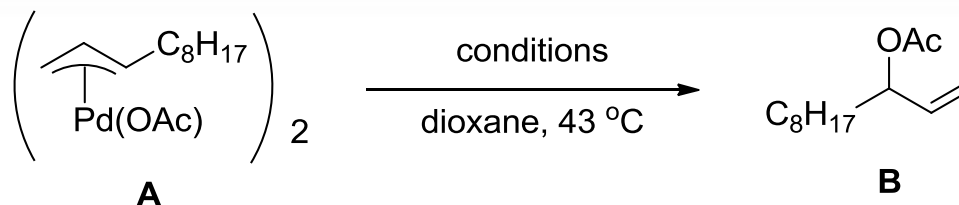
Mechanism Details



Khlar, N.; Araujo, C.; Alcudia, F.; Fernandez, I. *J. Org. Chem.* **2002**, 67, 345.



Exploration of Functionalization Step



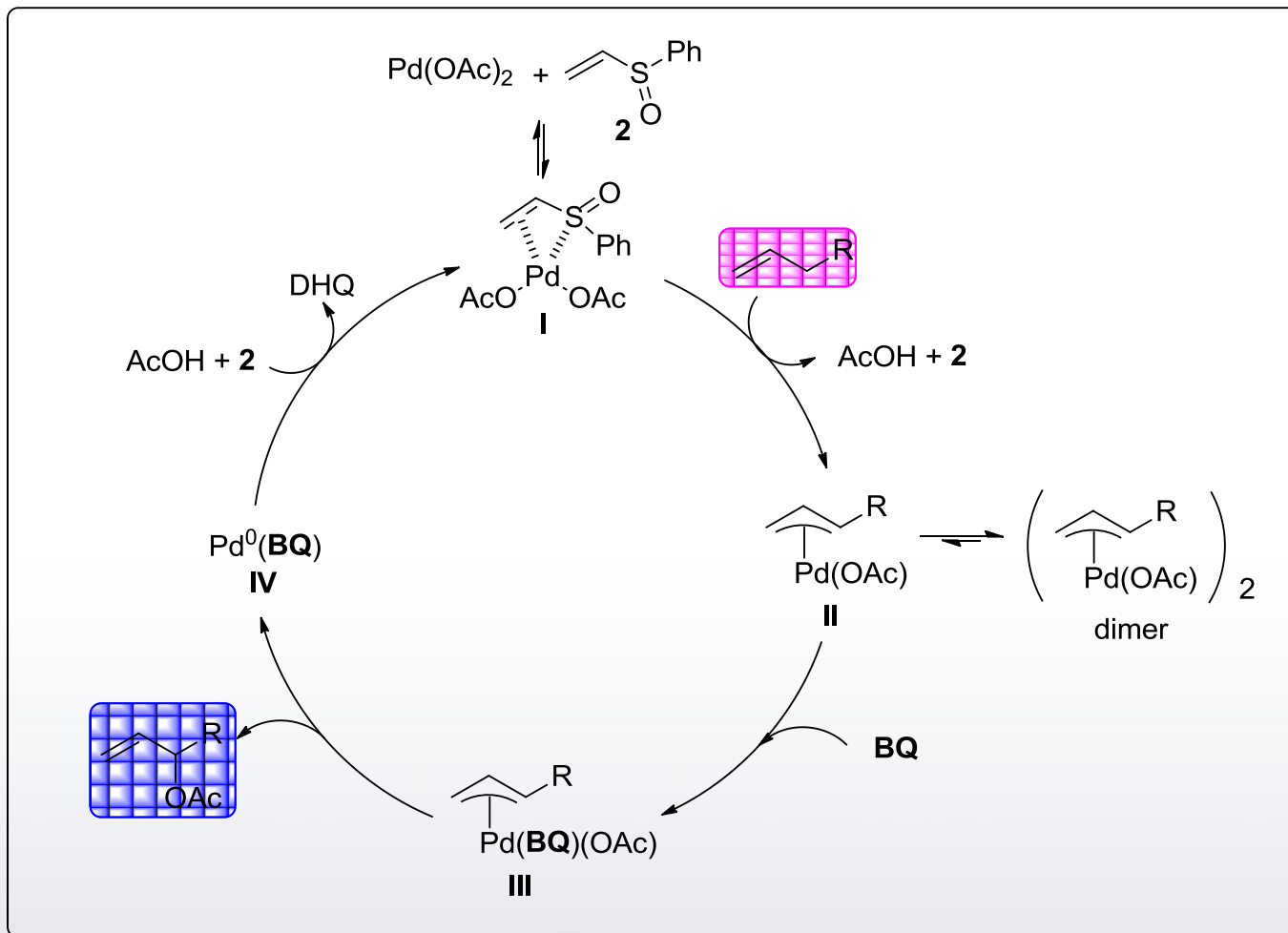
entry	conditions	yield (GC)	[B:L]
1	AcOH (40 equiv)	<1%	-
2	2 (1 equiv)	<1%	-
3	BQ (20 equiv)	58%	32:1
4	2 (1 equiv) BQ (20 equiv)	68%	34:1
5	PPh ₃ (20 equiv)	42%	1:1
6	dppe (10 equiv)	44%	1:1

The role of benzoquinone (BQ):

- An oxidant of Pd(0) to give Pd(II)
- A π -acidic ligand to activate π -allyl-Pd toward reductive elimination
- Regenerates $^-$ OAc to activate allylic C-H bond *in situ*.

Proposed Mechanism

Serial Ligand Catalysis



Contents

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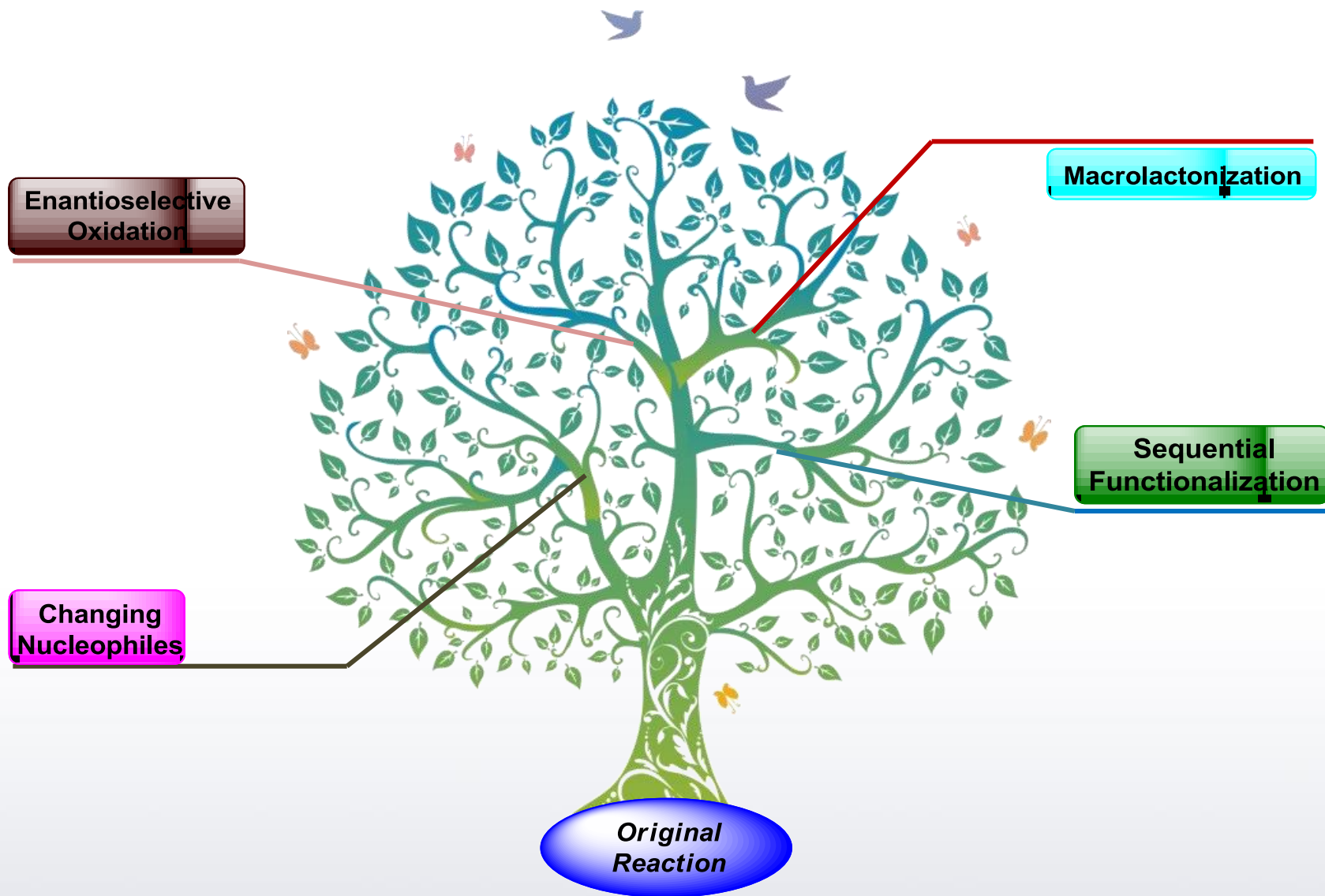
2 Original Reaction and Its Mechanism



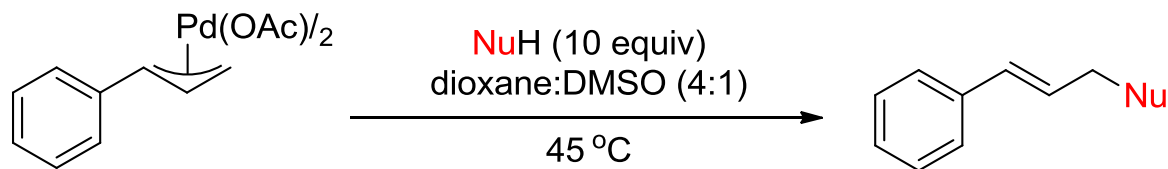
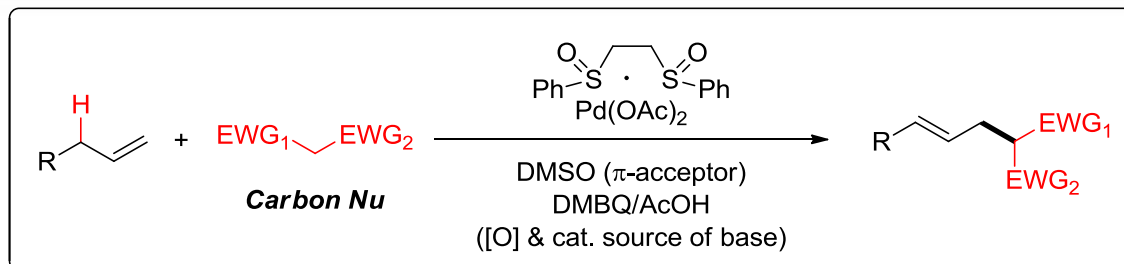
3 Reaction Evolution and Derivation

4 Synthetic Applications

Reaction Evolution and Derivation



Changing Nucleophiles



Acidity of nucleophiles is crucial for allylic C-H alkylation!

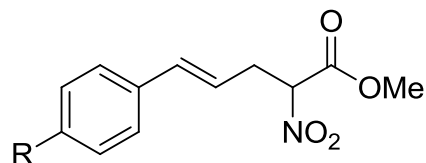
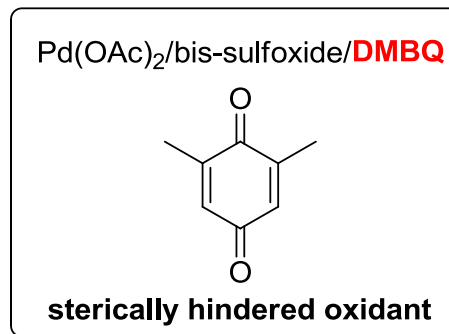
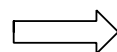
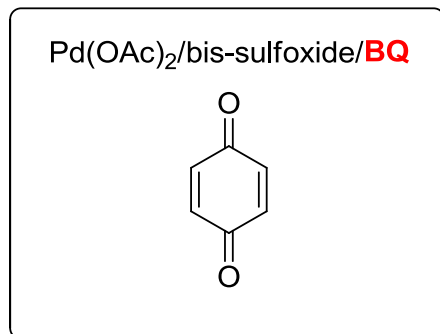


$\text{pK}_a < 6$

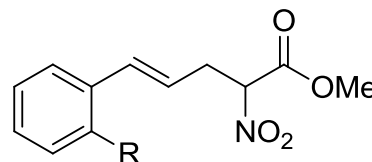
entry	NuH	yield (L+B)	L:B
1	PhO ₂ SCH ₂ CO ₂ Me, 3	9%	--
2	NO ₂ CH ₂ COPh, 4	82%	8:1
3	NO ₂ CH ₂ CO ₂ Me, 5	86%	4:1
4	NO ₂ CH ₂ SO ₂ Ph, 6	89%	16:1
5	5 (no DMSO)	--	--

Changing Nucleophiles

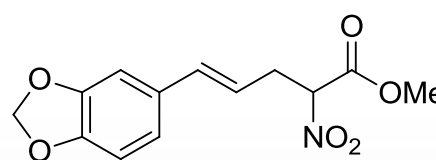
Select a suitable oxidant:



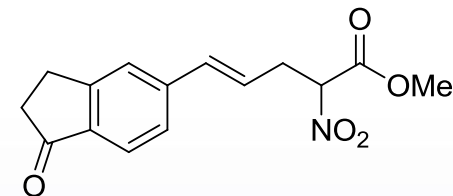
R = OMe,	50%,	[L:B] = 1.7:1
Me,	61%,	[L:B] = 3:1
CH ₂ =CH ₂ ,	63%,	[L:B] = 3:1
H,	62%,	[L:B] = 4:1
Br,	60%,	[L:B] = 4:1
NTsMe,	63%,	[L:B] = 4:1
F,	65%,	[L:B] = 4:1
CO ₂ Me,	61%,	[L:B] = 10:1
C(O)Me,	66%,	[L:B] = 10:1
CF ₃ ,	56%,	[L:B] = 10:1
CN,	65%,	[L:B] = 12:1



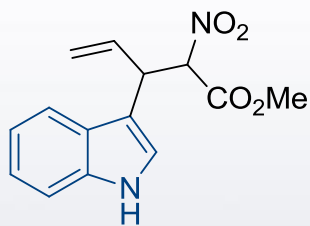
R = Me,	62%,	[L:B] = 5:1
CF ₃ ,	59%,	[L:B] > 20:1
OTf,	57%,	[L:B] = 15:1
OTBS,	58%,	[L:B] = 3:1



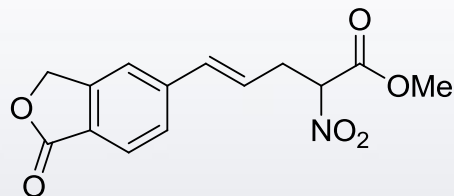
56%, [L:B] = 3:1



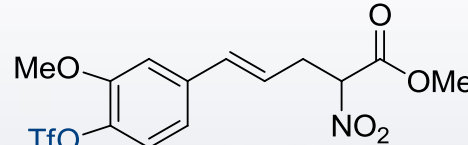
62%, [L:B] = 12:1



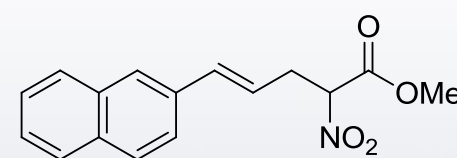
42%, [L:B] = 1:5



70%, [L:B] = 15:1

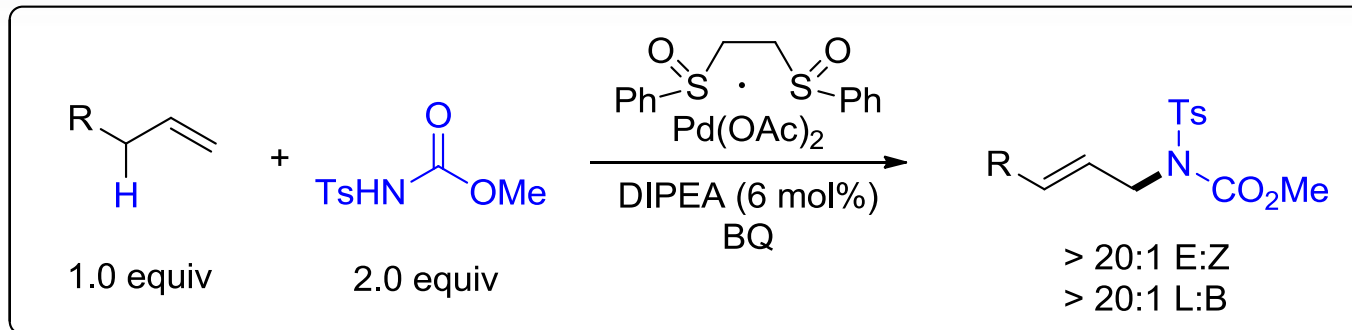


64%, [L:B] = 5:1

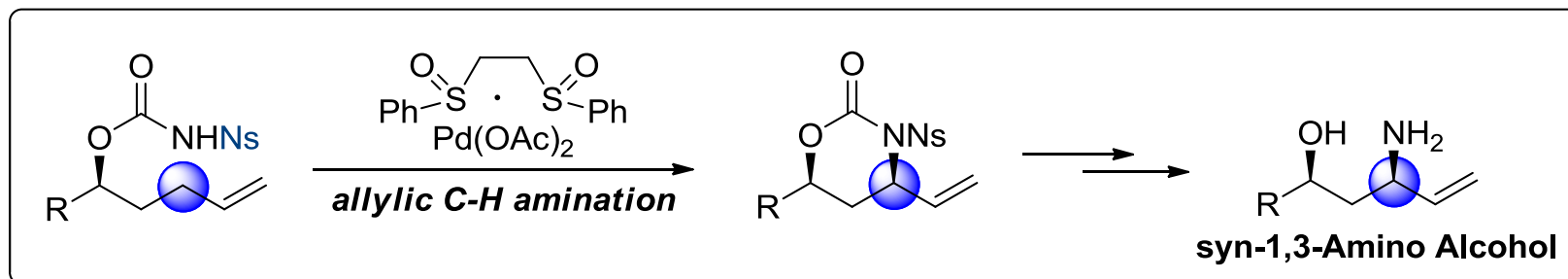


65%, [L:B] = 4:1

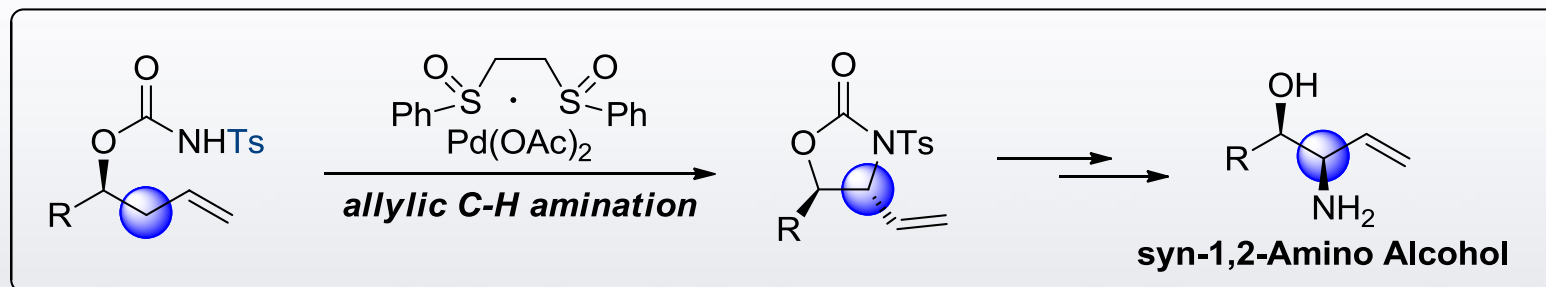
Changing Nucleophiles



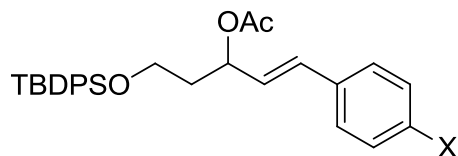
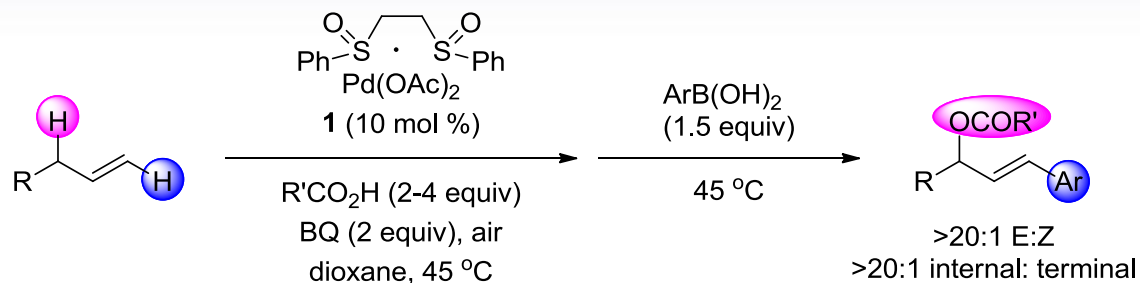
Reed, S. A.; Mazzotti, A. R.; White, M. C. *J. Am. Chem. Soc.* **2009**, *131*, 11701.



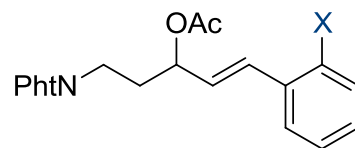
Rice, G. T.; White, M. C. *J. Am. Chem. Soc.* **2009**, *131*, 11707.



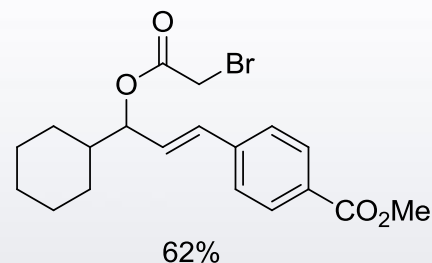
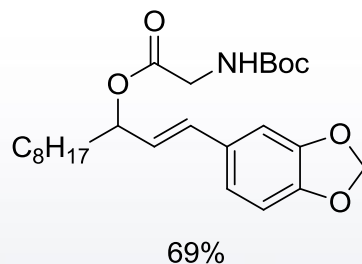
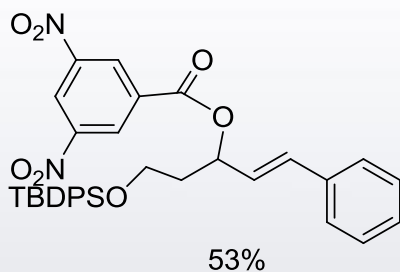
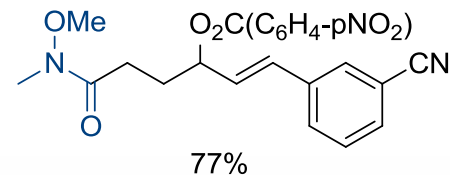
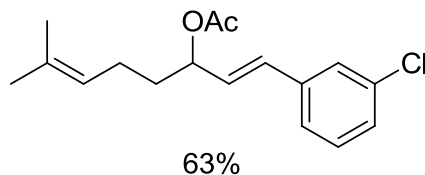
Sequential Functionalization



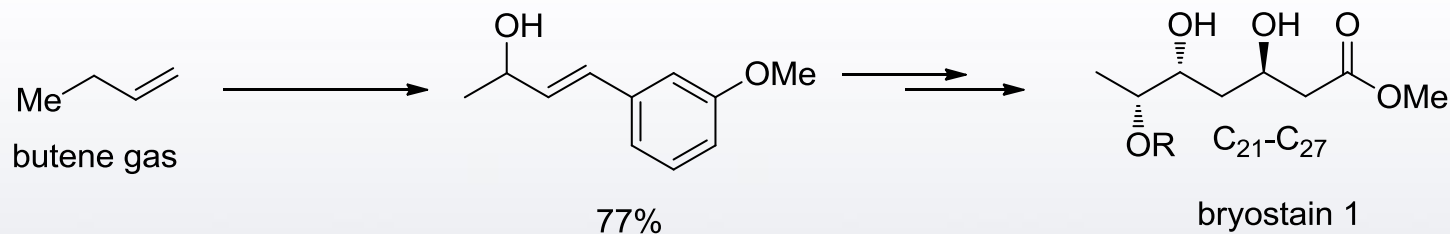
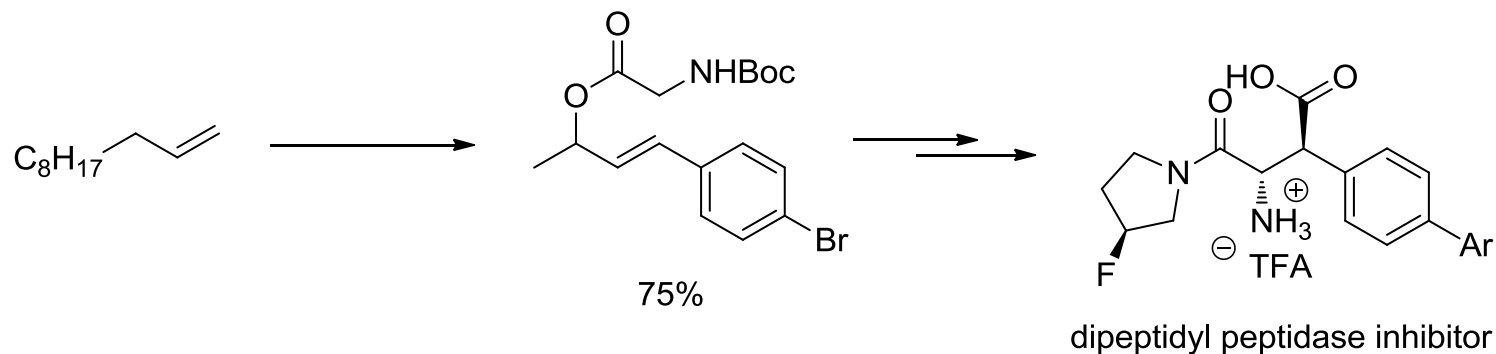
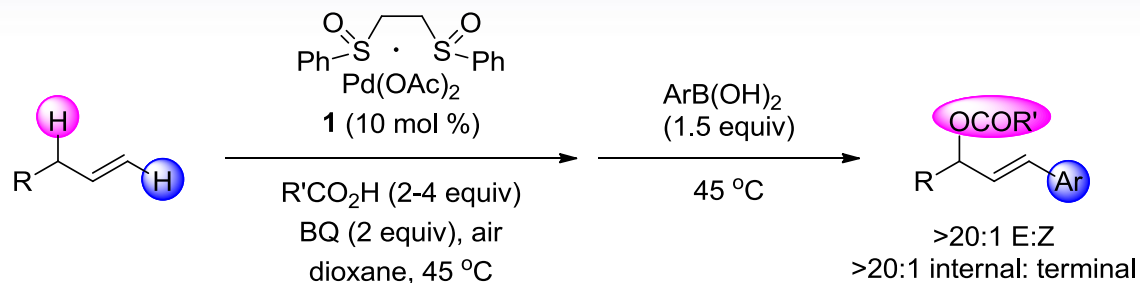
X = H, 74%
 X = OMe, 52%
 X = Cl, 63%
 X = F, 74%
 X = CHO, 60%



X = Me, 73%
 X = OMe, 55%

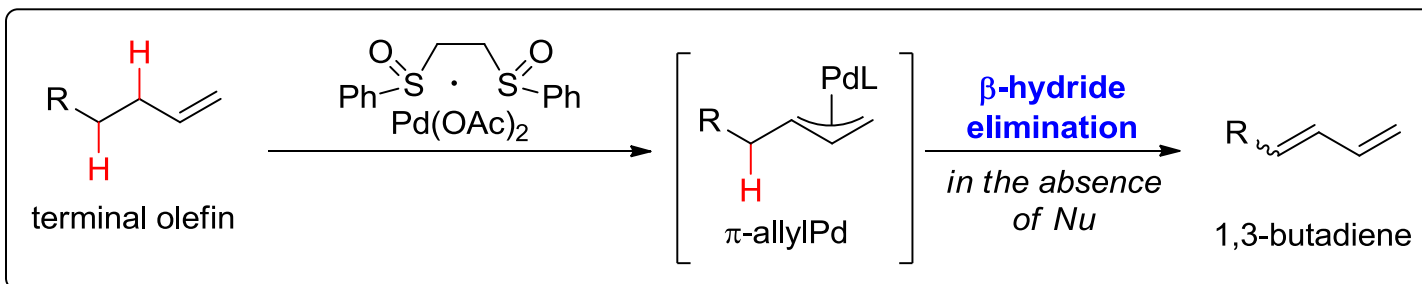


Sequential Functionalization



Sequential Functionalization

Dehydrogenation of Terminal Olefins ?



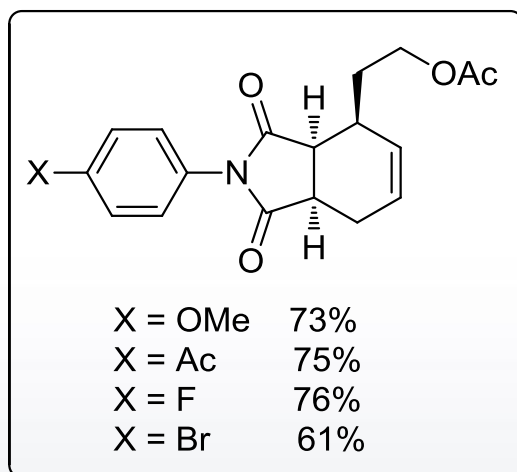
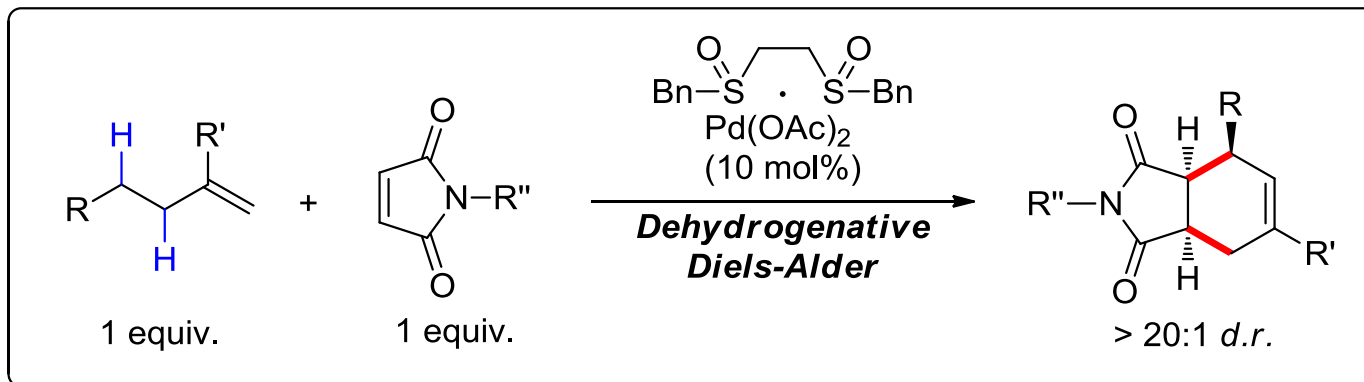
Challenge

- Dienes are prone to isomerization and oxidations.
- The electrophilic catalysts often catalyze diene oligomer- and polymerization processes.

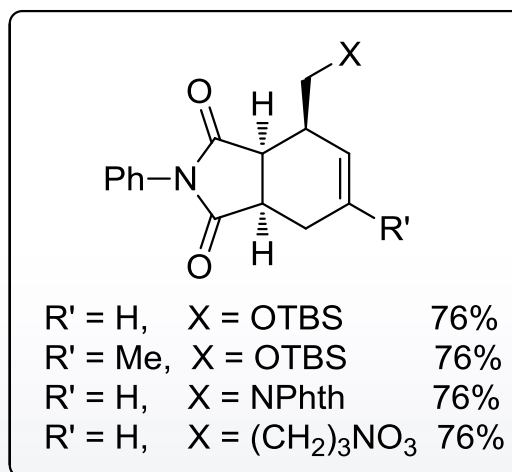
Strategy

Add a reactive component capable of furnishing a stable product

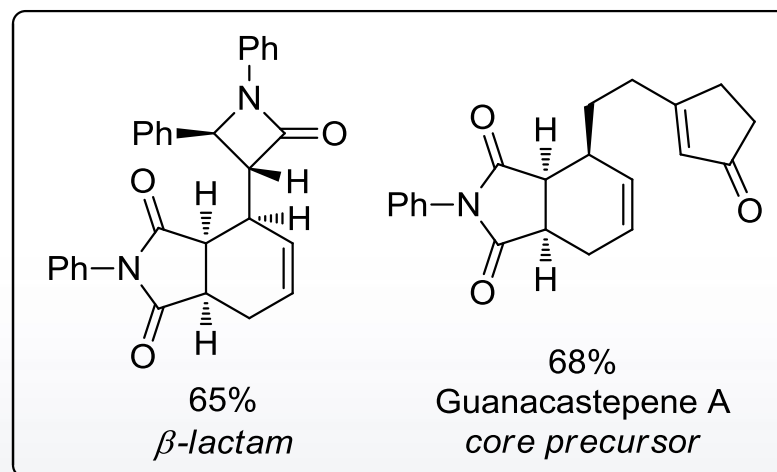
Sequential Functionalization



Dienophile Scope

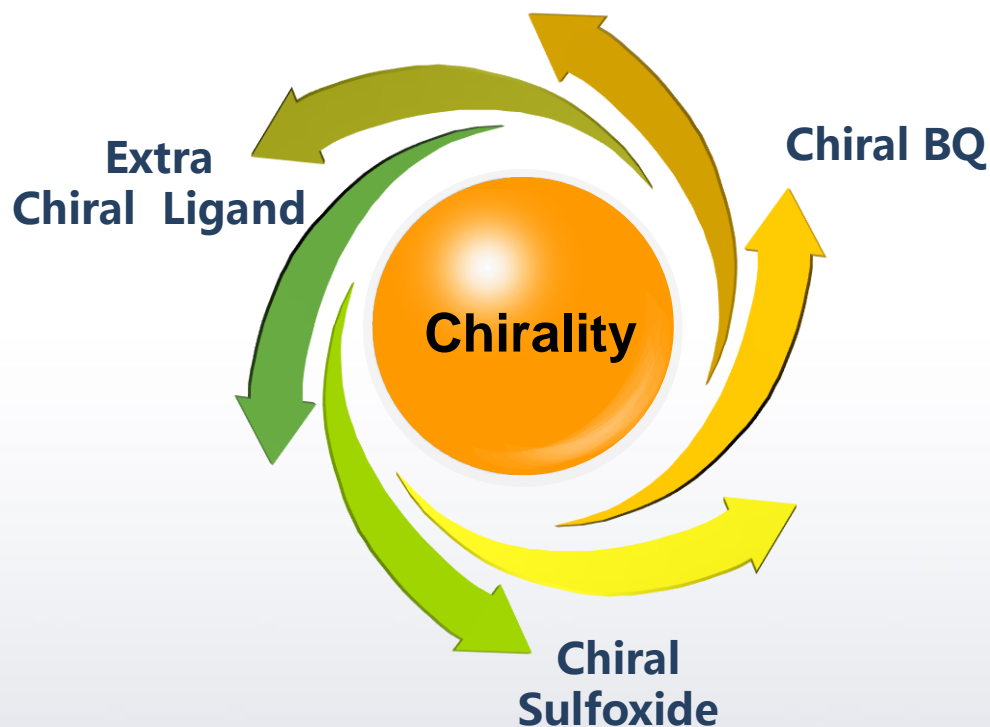
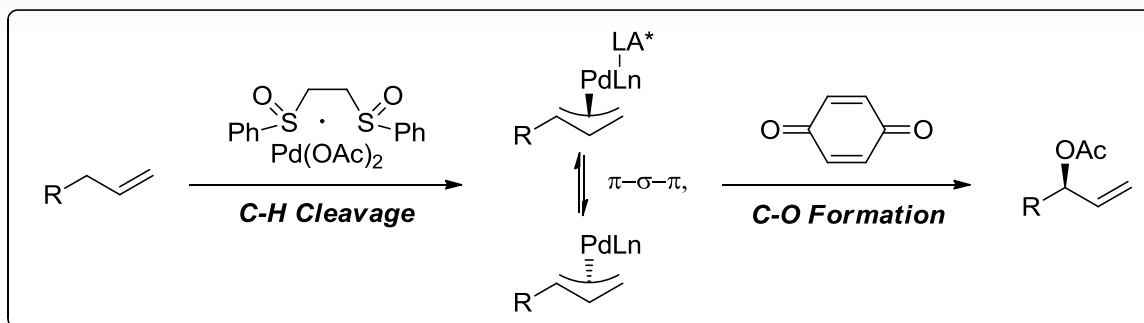


Olefin Scope



Medicinally Important Motifs

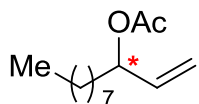
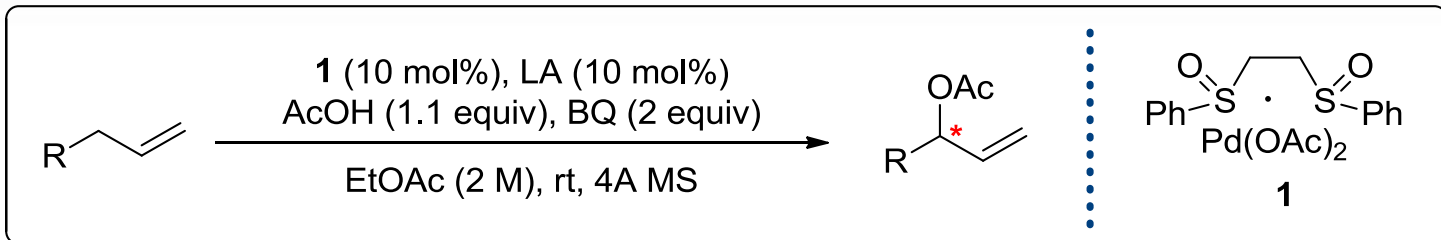
Enantioselective C-H Oxidation



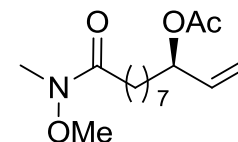
Challenge

- Reaction system can not tolerate strongly coordination ligands.
- Chiral sulfoxides were ineffective due to the rapid isomerization.
- Functionalization of BQ is impractical for covalent chiral modification,
- An oxophilic, chiral LA would increase the acid of BQ, and transmitt chiral information to the metal center.

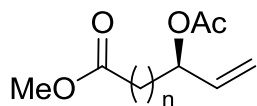
Enantioselective C-H Oxidation



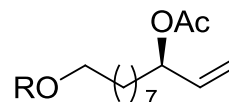
(R,R)-(salen)Cr^{III}F, 92%, ee 59%, [B:L] = 5.3:1
 (S,S)-(salen)Cr^{III}F, 92%, ee 59%, [B:L] = 5.3:1



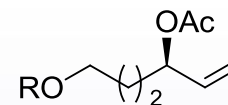
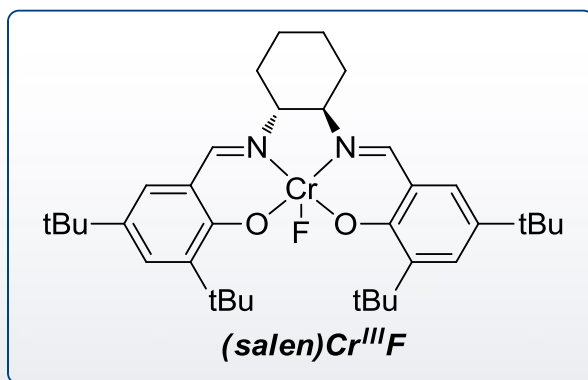
81%, ee 54%, [B:L] = 4.4:1



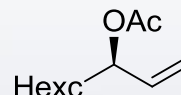
n = 7, 89%, ee 57%, [B:L] = 4.8:1
 n = 2, 69%, ee 50%, [B:L] = 4.6:1



R = TBDPS, 84%, ee 63%, [B:L] = 4.4:1
 R = H, 83%, ee 50%, [B:L] = 4.4:1



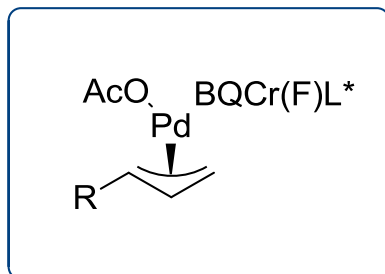
R = THP, 94%, ee 49%, [B:L] = 3.6:1
 R = Bn, 90%, ee 45%, [B:L] = 4.3:1



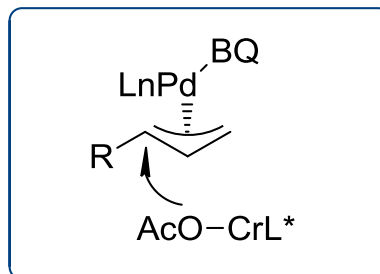
78%, ee 62%, [B:L] = 1.5:1

Enantioselective C-H Oxidation

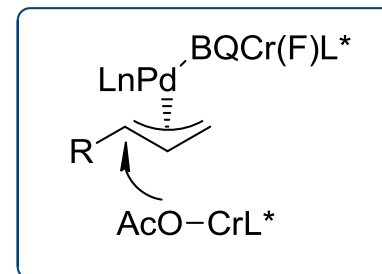
Three Mechanistic Scenarios



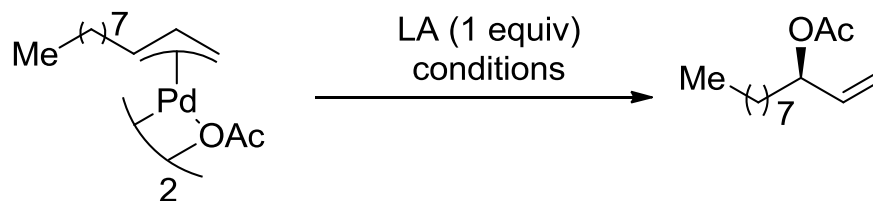
I



II



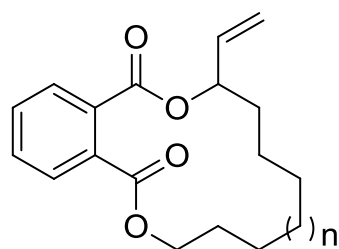
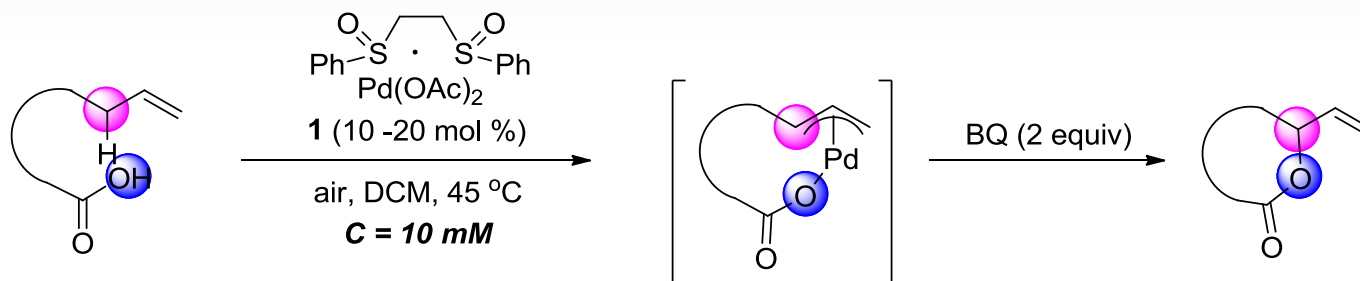
III



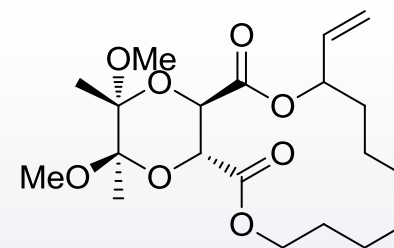
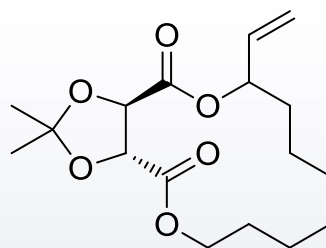
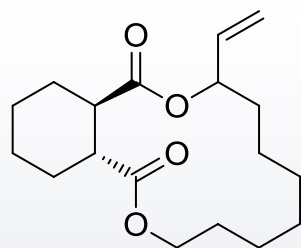
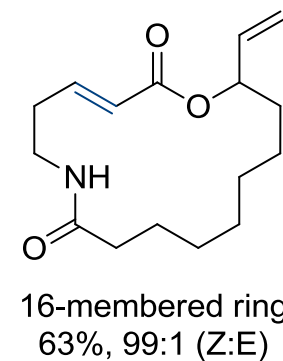
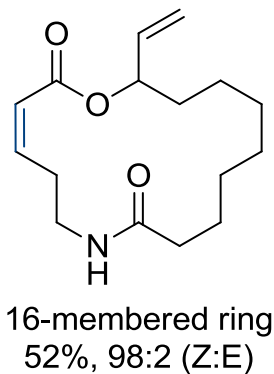
Entry	LA	K _{rel}	Yield	B:L	ee
1	--	1.0	20%	>20:1	--
2	(R,R)-3	9.7	85%	5.2:1	55%
3 ^a	(R,R)-3	--	0	--	--
4	(R,R)-4 ^b	3.8	41%	2.2:1	29%

^a No BQ added. ^b (R,R)-salen-Cr(OAc).

Macrolactonization

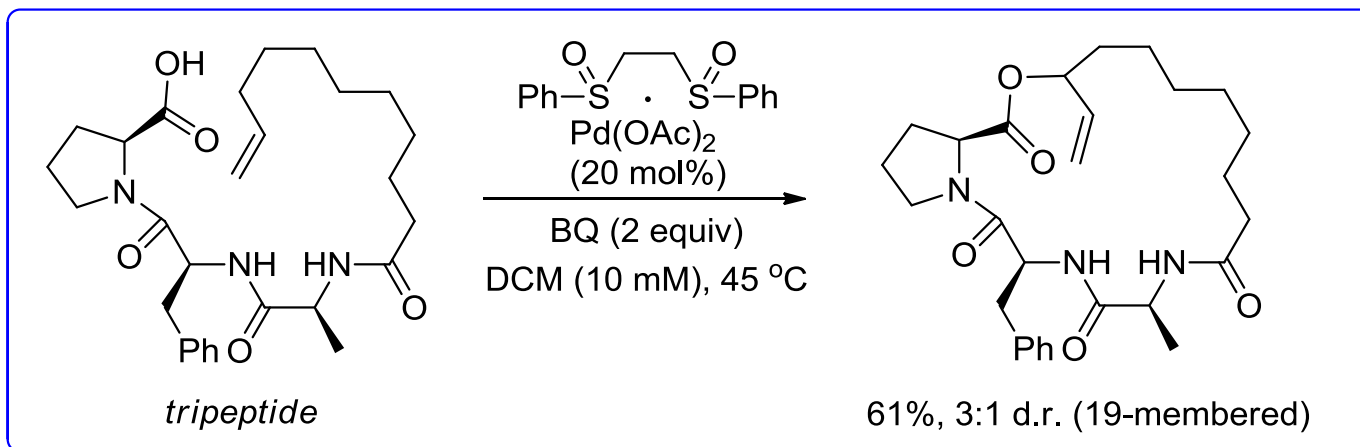


$n = 1, 61\%$
1 g scale, 62%
 $n = 2, 52\%$
 $n = 3, 60\%$
 $n = 4, 53\%$

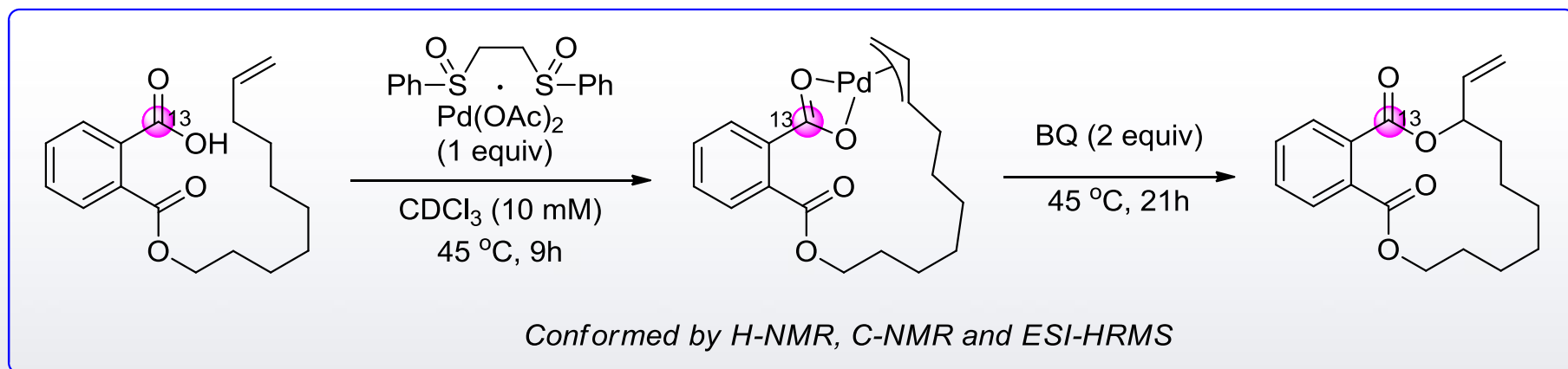


Macrolactonization

Macrocyclic Depsipeptide Synthesis



¹³C-Labeled Experiment



Contents

1 Introduction

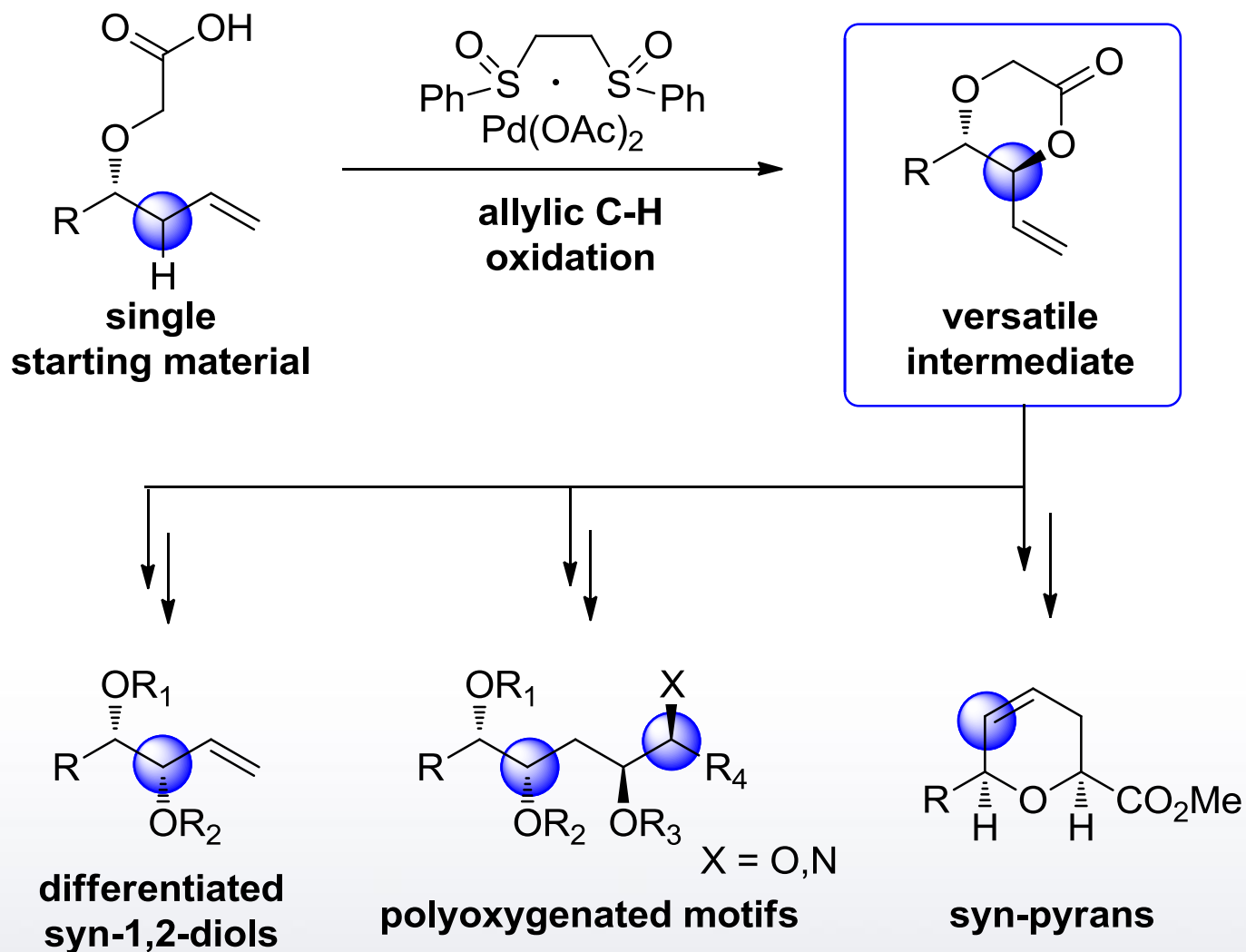
2 Original Reaction and Its Mechanism

3 Reaction Evolution and Derivation

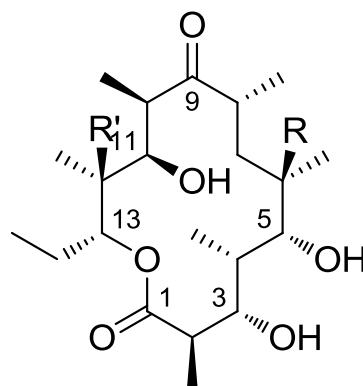
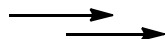
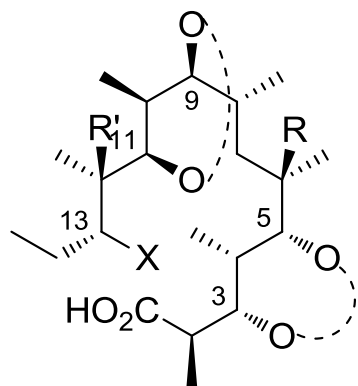


4 Synthetic Applications

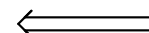
Rapid access to Complex Structures



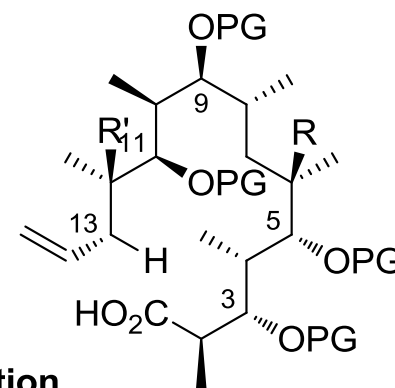
Natural Product Synthesis



Erythronolide A, R=R'=OH
Erythronolide B, R=OH, R'=H



**allylic C-H
Functionalization**

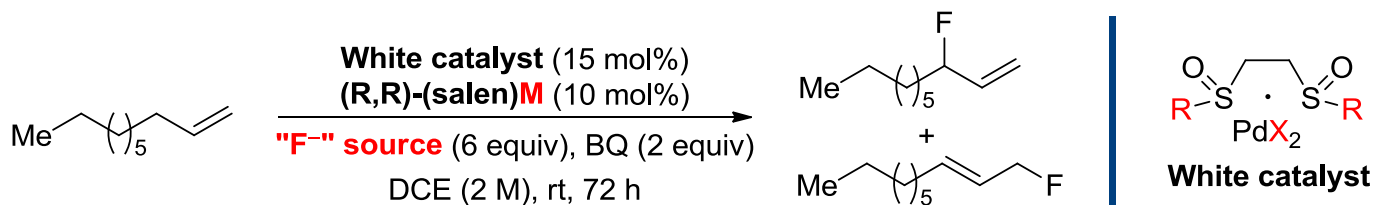
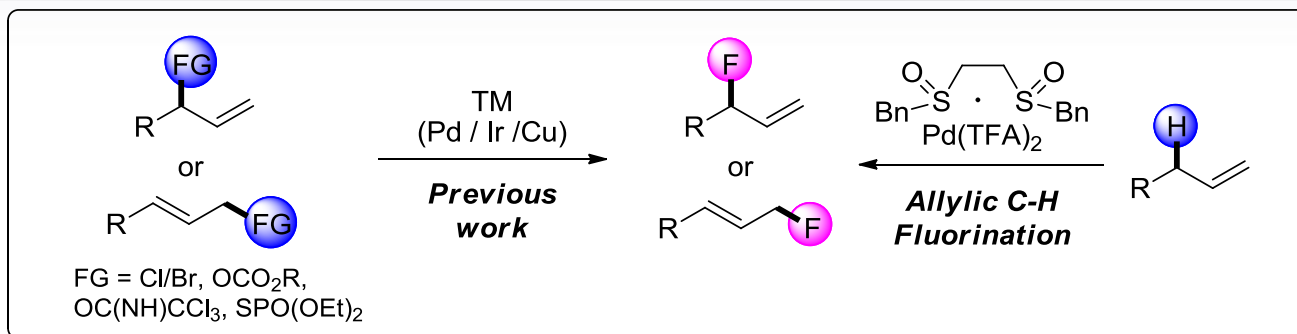


Woodward (1981)
Stork (1987)
Nakata (1989)
Danishefsky (1990)
Hoffmann (1993)
Evans (1998)
White (2009)
Nelson (2010)

"certain structural features such as ..cyclic protecting groups at C-3/C-5 and C-9/C-11 are required for efficient lactonization"

Preorganization is not required, although such elements can significantly improve the diastereomeric outcome of the cyclization.

Allylic C-H Fluorination



Entry	"F-" source	M	R	X	yield (%)	[B:L]
1	AgF	–	Ph	OAc	0	–
2	KF	–	Ph	OAc	0	–
3	KHF ₂	–	Ph	OAc	0	–
4	Py*9HF	–	Ph	OAc	0	–
5	TEA*3HF	–	Ph	OAc	33	6.6:1
6	TEA*3HF	CrCl	Ph	OAc	51	7.0:1
7	TEA*3HF	CoCl	Ph	OAc	6	7.9:1
8	TEA*3HF	MnCl	Ph	OAc	14	7.2:1
9	TEA*3HF	CrF	Ph	OAc	28	5.3:1
10	TEA*3HF	CrCl	Bn	OAc	65	6.0:1
11	TEA*3HF	CrCl	Bn	TFA	70	7.3:1

Allylic C-H Fluorination

